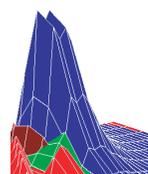
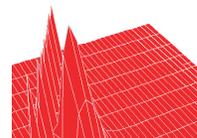
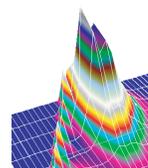
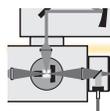


## Measurement of Environmental Contaminant Hydrocarbons by Fluorescence Spectroscopy

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Polycyclic aromatic hydrocarbons (PAH) and related chemicals are the largest known class of mutagens and carcinogens. They are frequent components of environmental contamination in water and soil. Contamination levels often occur in the ppm range, well above current EPA standards for acceptable levels, which are in the ppb range. This note describes the novel combined use of standard UV spectroscopy and fluorescence excitation-emission spectroscopy to measure low-ppm levels of PAH and other hydrocarbons. Standard Roper Scientific™/Acton Research spectroscopy components were used in a unique single system for both types of measurements. The results demonstrate that the two methods are complementary and that analysis of fluorescence excitation-emission matrices (EEM) of environmental contaminant hydrocarbons can sometimes provide greater sensitivity than UV spectroscopy.





## Experimental Methodology

The work reported in this note was conducted by James Brasch of JB Labs in Columbus, Ohio for Acton Research. The instrumentation consisted of a 75-W xenon source, a filter wheel, and Acton Research SpectraPro<sup>®</sup> spectrometers. A universal sample chamber with 0-degree and 90-degree exit ports permitted both the absorption and fluorescence measurements. A source-monitoring detector mounted at the monochromator exit at the 0-degree port was used to compensate for variations in the xenon source at each measurement. For the absorption studies, an Acton Research SpectraPro 150 served to illuminate the samples and a silicon detector in the 0-degree port of the sample chamber was employed. For the fluorescence measurements, the aforementioned SpectraPro 150 served as the excitation monochromator and a second SpectraPro 150 with a photon-counting detector was attached to the 90-degree port as the emission monochromator (Figure 1). To control the system and acquire data from the detectors, an Acton Research NCL<sup>™</sup> acquisition package and SpectraSense<sup>™</sup> software were utilized. The EEM were evaluated using Grams<sup>®</sup> 5.0 and 3-D software (Galactic Industries Corp.).

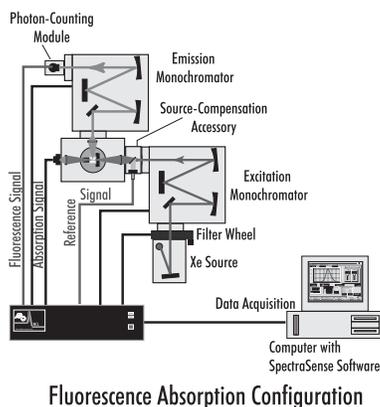


Figure 1. Experimental setup.

Three PAH were used in the study — anthracene, naphthalene, and pyrene. The non-PAH, stilbene, was also included because of its intense fluorescence spectrum. Ethanol stock solutions of the individual chemicals and mixtures of all four were prepared at 2-ppm and 20-ppm concentrations. All samples were subjected to both

absorption and fluorescence measurements to allow comparison of the techniques for low-concentration samples. Absorption measurements were collected in 2-nm increments at wavelengths between 200 nm and 660 nm using an integration time of 50 msec. Fluorescence readings were taken with excitation in 20-nm increments between 220 nm and 400 nm and emissions collected in 10-nm increments between 250 nm and 600 nm. EEM acquisition time was approximately 4 min using a 2-nm bandpass.

## UV and Fluorescence Measurement of Hydrocarbons

The four chemicals each had a distinct UV-absorbance spectrum at 20-ppm concentration (see Figure 2). However, Figure 3 shows that at 0.2-ppm concentrations, the three PAH were barely distinguishable and stilbene could not be visualized, even with first and second derivatives of the spectra.

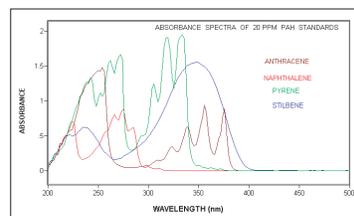


Figure 2

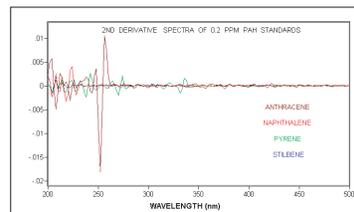
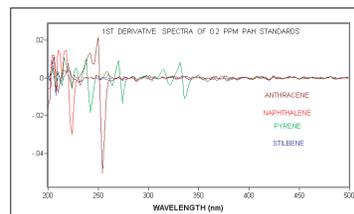


Figure 3

Figure 2. Absorbance spectra of 20-ppm samples.

Figure 3. Absorbance spectra first derivative (top tracing) and second derivative (bottom tracing) of 0.2-ppm samples.

With fluorescence measurements in which EEM were created, each of the four chemicals had a distinct EEM at the 2-ppm level (refer to Figures 4 - 7). Figure 8 shows a combined overlay of the EEM for all four chemicals (0.2 ppm) on the same intensity scale. By this method, there was a distinctive characteristic of each component that could be used for identification. In Figure 9, an EEM is shown for a 0.05-ppm mixture of all four hydrocarbons. All except pyrene were readily discernible at 0.05 ppm by the fluorescence methodology using two Acton Research SpectraPro 150 instruments.

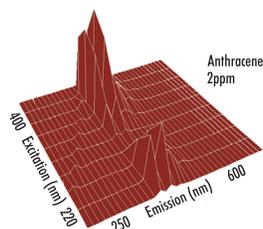


Figure 4

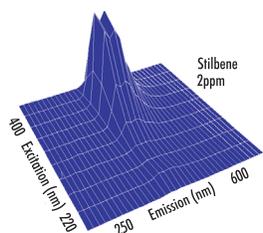


Figure 5

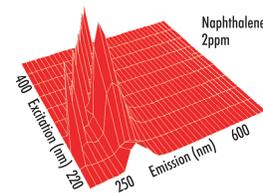


Figure 6

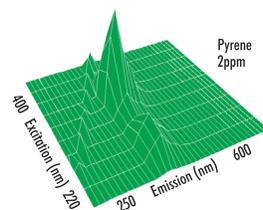


Figure 7

Figure 4. EEM of 2 ppm anthracene.  
 Figure 5. EEM of 2 ppm stilbene.  
 Figure 6. EEM of 2 ppm naphthalene.  
 Figure 7. EEM of 2 ppm pyrene.

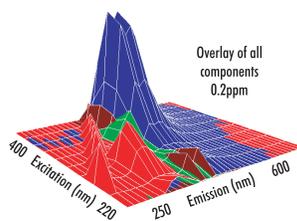


Figure 8

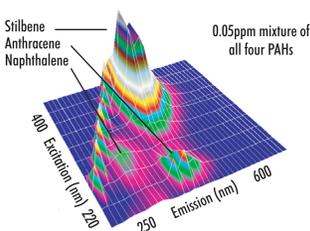


Figure 9

Figure 8. Combined overlay of EEM for all four chemicals at 0.2 ppm.  
 Figure 9. EEM of mixture of all four chemicals at 0.05 ppm.

In conclusion, fluorescence spectroscopy with EEM construction and analysis is a sensitive method that complements standard UV spectroscopy for low concentrations of hydrocarbons. Furthermore, this method of fluorescence spectroscopy can detect some chemicals at concentrations that are below the detection limits of standard UV absorption spectroscopy (e.g., stilbene at 0.2 ppm). This method could greatly facilitate the detection and monitoring of environmental contaminant PAH and other hydrocarbons.



## SpectraPro Monochromators and Spectrographs

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Roper Scientific offers a full line of accessories for SpectraPro instruments. These include light-input devices, light sources, universal sample chamber and sample holders, filter wheel assemblies, single-channel detector assemblies, and fiberoptic bundles and adapters. Spectroscopy components and accessories from Roper Scientific are the solutions for your Raman, fluorescence, emission, or absorption application needs.

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